

## 1. NAME OF THE MEDICINAL PRODUCT

Emzoron Capsules

## 2. QUALITATIVE AND QUANTITATIVE COMPOSITION

Each capsule contains FERROUS FUMARATE 45.3 EQ TO ELEMENTAL IRON 14.5, FOLIC ACID 0.3MG, VIT B12 3MCG, VIT B1 3MG, VIT B6 2MG, COPPER SULPHATE 2.51MG, DICALCIUM PHOSPHATE 339.5MG, ZINC SULPHATE 24.7MG EQ TO 10MG

### Excipients with known effect

Each capsule contains VIT B1 3MG, VIT B6 2MG, COPPER SULPHATE 2.51MG, DICALCIUM PHOSPHATE 339.5MG, ZINC SULPHATE 24.7MG EQ TO 10MG

For the full list of excipients, see section 6.1.

## 3. PHARMACEUTICAL FORM

Capsule, hard

Pink hard capsules containing 4 reddish-brown mini-tablets, 1 yellow mini-tablet and 3 white mini-tablets. The size of the capsule is 'Size 0' (dimensions 21.7x7.5mm).

## 4. CLINICAL PARTICULARS

### 4.1 Therapeutic indications

Treatment of iron deficiency with a proven deficiency of folic acid, especially one month before conception and until the third month of gestation.

### 4.2 Posology and method of administration

#### Posology

1 capsule once a day.

#### Method of administration

The capsules should be swallowed whole with plenty of liquid before breakfast or meals. In case of persi stingside effects, Emzoron Capsule can be taken with meals.

The duration of treatment depends on severity of the iron deficiency.

#### *Paediatric population*

Emzoron Capsule is not intended for use in children.

### 4.3 Contraindications

- Hypersensitivity to the active substances or to any of the excipients listed in section 6.1
- Clinical manifestations of iron accumulation: haemochromatosis, haemosiderosis
- Iron metabolism disorders
- Anaemia not a result of iron or folic acid deficiency
- Gastrointestinal disorders

### 2.1 Special warnings and precautions for use

In most cases, the treatment of iron deficiency anaemia leads to good therapeutic outcomes. However, it is particularly important to clarify the cause for iron deficiency anaemia.

Iron therapy should be continued after the haemoglobin has normalized (anaemia is defined as haemoglobin less than 11 g/dl in pregnant and less than 12 g/dl in non-pregnant women) in order to replenish iron stores (usually 4-12 weeks). Iron may lead to a black colouring of the stool. This does not have any clinical relevance.

The metabolism of folic acid and the metabolism of vitamin B<sub>12</sub> are intimately linked such that deficiency of either vitamin leads to an identical megaloblastic anaemia. Neurologic manifestations of folate deficiency overlap with those of vitamin B<sub>12</sub> deficiency and include cognitive impairment, dementia, depression, and, less commonly, peripheral neuropathy and subacute combined degeneration of the spinal cord. The inappropriate administration of folic acid in the presence of vitamin B<sub>12</sub> deficiency may lead to both neurologic and, later, hematologic relapse.

As ascorbic acid has an effect on stone formation, patients with nephrolithiasis or urolithiasis should use caution when taking <Invented name>. In patients with severe or terminal renal insufficiency (dialysis patients) a daily dose of 50 to 100 mg ascorbic acid should not be exceeded due to risk of hypoxalaemia and crystallisation of oxalate in the kidneys.

Ascorbic acid, a strong reducing agent, interferes with laboratory tests involving oxidation and reduction reactions. Falsely elevated or false-negative test results may be obtained from plasma, faeces, or urine samples depending on such factors as the dose of ascorbic acid and specific method used.

This medicinal product contains lactose. Patients with rare hereditary problems of galactose intolerance, the Lapp lactase deficiency or glucose-galactose malabsorption should not take this medicine.

## 2.2 Interaction with other medicinal products and other forms of interaction

Concomitant application of Emzoron Capsule with the following substances can influence therapeutic efficacy:

Interaction with	Possible effect
<i>Acetohydroxamic acid</i>	Reduced absorption of acetohydroxamic acid (chelator) and iron
<i>Acetylsalicylic acid, salicylates, NSAR, oxphenbutazon, phenylbutazon</i>	Enhanced gastrointestinal irritation
<i>Antacids (aluminium, calcium salts), calcium and magnesium preparations, magnesium hydroxide carbonate, magnesium carbonate, caffeine (coffee and tea), milk and milk products, sodium hydrogen carbonate, nutritional components (substances containing phytates, oxalates and phosphates), anion exchanger, polystyrene sulfonate</i>	Reduced plasma level of iron
<i>Anticonvulsives (e.g. carbamazepine, phenobarbital, primidone, valproic acid)</i>	Increasing folic acid ingestion can increase the activity of hepatic microsomal enzymes and thus the clearance of the anticonvulsant medications
<i>Antituberculous drugs, alcohol, glucarpidase</i>	Reduced plasma level of folate
<i>Auranofin</i>	Reduced action of auranofin
<i>Bisphosphonates</i>	Reduced plasma level of bisphosphonates

Interaction with	Possible effect
<i>Cefdinir</i>	Reduced absorption of cefdinir
<i>Chloramphenicol</i>	Reduced action of folate, avoid combination
<i>Cimetidine</i>	Reduced plasma level of iron
<i>Dapsone</i>	Reduced plasma level of iron due to reduced absorption (chelator)
<i>Entacapone</i>	Reduced plasma level of entacapone (chelator) and iron
<i>Folic acid antagonists (e.g. methotrexate, pyrimethamine, triamterene, trimethoprim, sulfonamides)</i>	Reduced plasma level of folate
<i>Levodopa, methyl dopa</i>	Reduced action of levodopa and methyl dopa due to reduced plasma levels
<i>Lipoic acid</i>	Combination with iron should be avoided if lipoic acid is administered orally (chelator)
<i>Mycophenolate mofetil</i>	Reduced plasma level of mycophenolate mofetil
<i>Omeprazole</i>	Affected bioavailability of dietary ascorbic acid
<i>Pancreatin and analogues</i>	Reduced plasma level of folate
<i>Penicillamine</i>	Reduced plasma level of penicillamine (chelator) and iron
<i>Primidone</i>	Reduced plasma level of folate
<i>Quinolones (e.g. ciprofloxacin, levofloxacin, moxifloxacin, ofloxacin, piperimidic acid)</i>	Iron salts affect the absorption of Quinolones
<i>Sulfasalazine</i>	Mutual reduction of action and inhibition of absorption (chelator)
<i>Tannin protein</i>	Reduced action of iron
<i>Tetracycline</i>	Reduced plasma level of tetracycline and iron
<i>Thyroid hormones, thyroxine</i>	Reduced plasma level of thyroid hormones and thyroxine
<i>Tilactase (Aspergillus oryzae)</i>	Reduced action of tilactase
<i>Tiopronin, dimercaprol</i>	Reduced action of iron, avoid combination (chelator); together with iron dimercaprol may form toxic complexes
<i>Trientine (chelating agent)</i>	Reduced absorption of iron, probably because it chelates with iron in the gut. A separation of at least 2 hours is recommended
<i>Zinc salts</i>	Reduced absorption of iron

### 2.3 Fertility, pregnancy and lactation

#### Pregnancy

Emzoron Capsule is indicated for the administration during pregnancy.

For pregnant women the daily administration of folate with iron is beneficial since anemia during pregnancy is usually caused by a deficiency of both nutrients.

Folic acid deficiency, the most common cause of megaloblastic anemia in pregnancy, is associated with open neural tube defects and other complications. Brea

#### st-feeding

Iron, folic acid and ascorbic acid are excreted into breast milk. This must be taken into consideration if the infant is receiving any respective supplements.

#### Fertility

There are no data on the effect of Emzoron Capsule on fertility.

### **2.4 Effect on ability to drive and use machines**

Emzoron Capsule has no or negligible influence on the ability to drive and use machines.

### **2.5 Undesirable effects**

#### Summary of the safety profile

Adverse reactions of supplemental iron such as heartburn, nausea, abdominal discomfort, constipation, and diarrhoea are dose related. These side effects may already occur for the first few days of therapy, and then they can subside.

With folic acid, hypersensitivity reactions have been reported spontaneously.

Large doses of ascorbic acid are reported to cause diarrhoea and other gastrointestinal disturbances. It has also been stated that large doses may result in hyperoxaluria and the formation of renal calcium oxalate calculi (see section 4.9).

#### Tabulated list of adverse reactions related to the active substances of the medicinal product

	<b>Frequency not known (cannot be estimated from the available data)</b>	
	<b>Oral Ferrous fumarate</b>	<b>Oral Folic acid</b>
<i>Immune system disorders</i>	hypersensitivity reactions including anaphylaxis	hypersensitivity reactions including anaphylactic reaction
<i>Metabolism and nutrition disorders</i>	haemosiderosis/haemochromatosis (in case of iron overload)	
<i>Gastrointestinal disorders</i>	gastrointestinal irritation and abdominal pain with nausea and vomiting, diarrhoea, constipation, black coloured faeces, heartburn	
<i>Skin and subcutaneous disorders</i>	hypersensitivity reactions such as rash, pruritus, urticaria, erythema, oedema, photosensitivity	hypersensitivity reactions such as rash, pruritus, urticaria

#### Reporting of suspected adverse reactions

Reporting suspected adverse reactions after authorization of the medicinal product is important. It allows continued monitoring of the benefit/risk balance of the medicinal product. Healthcare professionals are asked to report any suspected adverse reactions via the national reporting system listed in Appendix V:

### **2.6 Overdose**

Acute iron toxicity occurs after ingestion of large quantities of iron salts. This can result in severe necrotizing gastritis with vomiting, haemorrhage and diarrhoea, followed by circulatory collapse. Increased capillary

permeability, reduced plasma volume, increased cardiac output, and sudden cardiovascular collapse may occur in acute iron intoxication.

Chronic iron toxicity or iron overload is virtually always due to causes other than ingestion of iron salts, most common being the giving of repeated blood transfusions to treat hemolytic anemia, notably the thalassemias.

The treatment of acute and chronic iron toxicity involves the use of iron chelators. The agent in common use is deferoxamine, which forms a complex with ferric iron. This complex is excreted in the urine. For further information on posology and method of administration, please refer to the Summary of Product Characteristics of deferoxamine.

Oral folic acid is usually not toxic. Even with doses as high as 15 mg per day, there have been no substantiated reports of side effects.

High or single doses of ascorbic acid, exceeding the tolerable upper intake level of 2,000 mg, can cause osmotic induced diarrhea and other gastrointestinal disturbances, haemolysis in patients with G6PD deficiency, hyperoxaluria, formation of renal calcium oxalate calculi.

### **3. PHARMACOLOGICAL PROPERTIES**

#### **3.1 Pharmacodynamic properties**

Pharmaco-therapeutic group: anti-acidemic preparations, iron in other combinations ATC code: B03AE10

Emzoron Capsule contains the active ingredients ferrous fumarate, folic acid and ascorbic acid in mini-tablets. The ferrous fumarate mini-tablets have a prolonged release formulation to prevent high local iron concentrations which can irritate the mucosa.

This preparation is suitable for the restoration of normal iron and folic acid levels in the blood during pregnancy, especially in the month before conception and until the third month of gestation. Thus, it prevents complications, e.g. anemia, abortions, hemorrhages, preterm deliveries and mental development disorders of the child due to iron and folic acid deficiency.

Supplementation with folic acid is recommended to prevent congenital malformations including neural tube defects. Neural tube defects develop in the first weeks after conception, a period during which pregnancy may not have been diagnosed yet, thus supplementation with folic acid is essential at the stage when pregnancy is being planned.

Ascorbic acid may stimulate iron absorption partly by forming soluble iron-ascorbate chelates and partly by reducing ferric iron ( $\text{Fe}^{3+}$ ) to the more soluble ferrous form ( $\text{Fe}^{2+}$ ). An enhanced absorption of the iron present in Emzoron Capsule by the contained ascorbic acid has not been demonstrated.

#### **3.2 Pharmacokinetic properties**

##### Absorption

- The amount of absorbed iron varies between 5 and 35%.
- Folic acid is absorbed mainly in the upper intestinal tract (duodenum, jejunum).
- Ascorbic acid is readily absorbed from the gastrointestinal tract.

##### Distribution

- Most absorbed iron is bound to transferrin and transported to the bone marrow where it is incorporated into hemoglobin; the remainder is contained within the storage forms, ferritin or

hemosiderin, or as myoglobin, with smaller amounts occurring in haem-containing enzymes or in plasma bound to transferrin.

- Once absorbed, folate is transported rapidly to tissues as  $\text{CH}_3\text{H}_4\text{PteGlu}$  (5-methyltetrahydrofolate). Although certain plasma proteins do bind folate derivatives, they have a greater affinity for non-methylated analogues.
- After absorption ascorbic acid is widely distributed in the body tissues. Plasma concentrations of ascorbic acid rise with increasing dose until a plateau is reached with doses of about 90 to 150 mg daily.

### Biotransformation and elimination

- Only very small amounts of iron are excreted as the majority released after the destruction of the hemoglobin molecule is re-used. This conservation of body iron, and lack of an excretory mechanism for excess iron, is the reason for the development of iron overload with excessive iron therapy or repeated transfusions.
- The liver actively reduces and methylates folic acid and di-ortetrahydrofolate and then transports the  $\text{CH}_3\text{H}_4\text{PteGlu}$  into bile for reabsorption by the gut and subsequent delivery to tissues. Folate metabolites are eliminated in the urine
- Ascorbic acid is reversibly oxidized to dehydroascorbic acid; some is metabolized to ascorbate-2-sulfate, which is inactive, and oxalic acid. Ascorbic acid and its metabolites are excreted in the urine.

### **3.3 Preclinical safety data**

Non-

clinical data reveal no special hazard for humans based on conventional studies of safety pharmacology, repeated dose toxicity, genotoxicity, carcinogenic potential, toxicity to reproduction and development.

## **4. PHARMACEUTICAL PARTICULARS**

### **4.1 List of excipients**

Lactose monohydrate  
Polyacrylate dispersion 30%  
Colloidal anhydrous silica  
Magnesium stearate  
Microcrystalline cellulose  
Stearic acid

Erythrosine (E127)  
Red iron oxide (E172)

Black iron oxide (E172)  
Titanium dioxide (E171)  
Gellan gum

### **4.2 Incompatibilities**

Not applicable.

### **4.3 Shelf life**

30 months

### **4.4 Special precautions for storage**

Do not store above 30°C.

### **4.5 Nature and content of container**

10, 20, 28, 30, 50, 56, 60, 84, 90, 98 and 100 capsules in PVC/aluminum blister packs. Not all pack sizes may be marketed.

**4.6 Specialprecautionsfordisposal**

Nospecialrequirements.

**5. MARKETINGAUTHORISATIONHOLDER**

Emzor pharmaceutical Industries, Plot 3c Block A aswani market road,  
Isolo-oshodi expressway Lagos

**6. MARKETINGAUTHORISATIONNUMBER(S)**

NA

**7. DATEOFFIRSTAUTHORISATION/RENEWALOFTHEAUTHORISATION**

NA

**8. DATEOFREVISIONOFTHETEXT**

NA